

Dicarbonyldi- μ -chloro-*cis,cis*- η^4 -1,5-cyclooctadienedirhodium(I)

Maurice Abou Rida,^a Joseph Saikaili,^b Anthony K. Smith^{a*} and Alain Thozet^c

^aLaboratoire de Chimie Organométallique de Surface, CNRS UMR 9986, Ecole Supérieure de Chimie Physique Electronique de Lyon, 43 Boulevard du 11 Novembre 1918, F-69616 Villeurbanne CEDEX, France, ^bFaculty of Science II, Lebanese University, Fanar Beirut, Lebanon, and ^cLaboratoire de Cristallographie, Université Claude Bernard Lyon 1, 43 Boulevard du 11 Novembre 1918, F-69622 Villeurbanne CEDEX, France
Correspondence e-mail: smith@cpe.fr

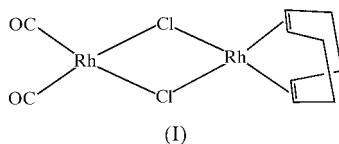
Received 6 October 2000

Accepted 11 December 2000

The title compound, dicarbonyl-1 κ^2 C-di- μ -chloro-1:2 κ^4 Cl-[*cis,cis*-2(η^4)-1,5-cyclooctadiene]dirhodium(I), [Rh₂Cl₂(C₈H₁₂)₂(CO)₂], consists of a dichloro-bridged dimer of rhodium, with a non-bonded Rh···Rh distance of 3.284 (2) Å. One Rh atom is coordinated to two carbonyl ligands, while the other Rh atom is coordinated to the cyclooctadiene moiety.

Comment

The crystal structures of the compounds [Rh₂Cl₂(CO)₄] (Dahl *et al.*, 1961; Walz & Scheer, 1991) and [Rh₂Cl₂(C₈H₁₂)₂] (Ibers & Snyder, 1962; Boeyens *et al.*, 1986; De Ridder & Imhoff, 1994) have been reported. However, the structure of the mixed-ligand rhodium dimer, with two carbonyl ligands coordinated to one Rh atom and a cyclooctadiene ligand coordinated to the other Rh atom, *i.e.* (I), has only been determined by X-ray powder diffraction (Corradi *et al.*, 1997).



The structure of (I) (Fig. 1) consists of two square-planar Rh atoms linked by two bridging Cl atoms. The bond lengths and angles of the Rh1 moiety are very similar to those found in the [Rh₂Cl₂(C₈H₁₂)₂] molecule, while those of the Rh2 moiety are similar to those found in the [Rh₂Cl₂(CO)₄] molecule. This is particularly remarkable with regard to the different Rh–Cl bond lengths. Thus, Rh1–Cl1 of 2.406 (2) Å and Rh1–Cl2 of 2.412 (2) Å correspond to the equivalent bond lengths found in [Rh₂Cl₂(C₈H₁₂)₂] (average 2.410 Å; De Ridder & Imhoff, 1994), while the shorter bond lengths, Rh2–Cl1 of 2.371 (2) Å and Rh2–Cl2 of 2.372 (2) Å, correspond to the equivalent bond lengths found in [Rh₂Cl₂(CO)₄] (average 2.384 Å; Walz & Scheer, 1991). As expected also, the Rh···Rh distance in

the title compound [3.284 (2) Å] approximates to the average of the corresponding values for [Rh₂Cl₂(C₈H₁₂)₂] [3.509 (1) Å; Boeyens *et al.*, 1986] and [Rh₂Cl₂(CO)₄] [3.138 (1) Å; Walz & Scheer, 1991]. The dihedral angle about the Cl···Cl vector of the Rh₂Cl₂ core is 139.6°, somewhat larger than that of [Rh₂Cl₂(CO)₄] (128.6°; Dahl *et al.*, 1961; Walz & Scheer, 1991), but markedly different from the planar Rh₂Cl₂ core in [Rh₂Cl₂(C₈H₁₂)₂] (Ibers & Snyder, 1962; Boeyens *et al.*, 1986; De Ridder & Imhoff, 1994). There are relatively short Rh···Rh intermolecular contacts (Rh2···Rh2' 3.50 Å) between the Rh atoms bearing the carbonyl ligands. The structure obtained from powder diffraction methods (Corradi *et al.*, 1997) showed a rather short packing contact between O1 and O1' [2.67 (6) Å] which was recognized to be susceptible to small structural changes. The present data show this contact to be much longer at 3.19 Å, otherwise there are no major differences in the structural parameters.

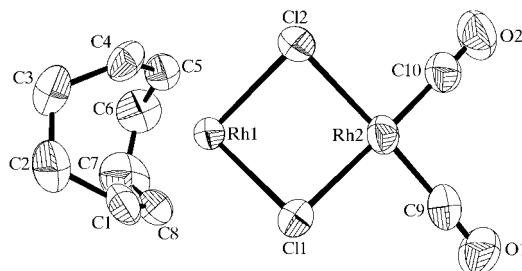


Figure 1

The molecular structure of the title compound with the atom numbering. Displacement ellipsoids are drawn at the 50% probability level.

Experimental

The title compound was extracted as a minor product from the solution resulting from the reaction between [RuCl(C₅H₅)(C₈H₁₂)] and [Rh₂Cl₂(CO)₄]. Orange crystals were grown from a CH₂Cl₂–hexane solution.

Crystal data

[Rh ₂ Cl ₂ (C ₈ H ₁₂) ₂ (CO) ₂]	$D_x = 2.238 \text{ Mg m}^{-3}$
$M_r = 440.92$	Mo $K\alpha$ radiation
Monoclinic, $P2_1/c$	Cell parameters from 15 389 reflections
$a = 6.6500 (12) \text{ \AA}$	$\theta = 2.09\text{--}30.02^\circ$
$b = 12.261 (3) \text{ \AA}$	$\mu = 2.912 \text{ mm}^{-1}$
$c = 16.058 (3) \text{ \AA}$	$T = 293 (2) \text{ K}$
$\beta = 92.19 (3)^\circ$	Block, orange
$V = 1308.3 (5) \text{ \AA}^3$	$0.15 \times 0.10 \times 0.10 \text{ mm}$
$Z = 4$	

Data collection

Nonius KappaCCD diffractometer	$\theta_{\min} = 2.09^\circ$
Area-detector scans	$\theta_{\max} = 30.02^\circ$
15 389 measured reflections	$h = 0 \rightarrow 9$
3699 independent reflections	$k = 0 \rightarrow 17$
2261 reflections with $I > 2\sigma(I)$	$l = -22 \rightarrow 22$
$R_{\text{int}} = 15389$	

Refinement

Refinement on F^2	H atoms constrained
$R[F^2 > 2\sigma(F^2)] = 0.067$	$w = 1/[σ^2(F_o^2) + (0.1P)^2]$
$wR(F^2) = 0.196$	where $P = (F_o^2 + 2F_c^2)/3$
$S = 1.118$	$(Δ/σ)_{\max} = 0.001$
3699 reflections	$Δρ_{\max} = 1.01 \text{ e \AA}^{-3}$
145 parameters	$Δρ_{\min} = -1.28 \text{ e \AA}^{-3}$

Table 1Selected geometric parameters (\AA , $^{\circ}$).

Rh1—C1	2.099 (8)	O2—C10	1.125 (10)
Rh1—C4	2.118 (7)	O1—C9	1.121 (10)
Rh1—C5	2.094 (7)	C1—C8	1.395 (14)
Rh1—C8	2.101 (8)	C1—C2	1.516 (12)
Rh1—Cl1	2.406 (2)	C2—C3	1.476 (12)
Rh1—Cl2	2.412 (2)	C3—C4	1.512 (12)
Rh2—C9	1.844 (9)	C4—C5	1.365 (11)
Rh2—C10	1.850 (9)	C5—C6	1.493 (13)
Rh2—Cl1	2.371 (2)	C6—C7	1.487 (13)
Rh2—Cl2	2.372 (2)	C7—C8	1.504 (14)
Cl1—Rh1—Cl2	85.03 (8)	C10—Rh2—Cl2	91.3 (3)
C9—Rh2—C10	91.4 (4)	Cl1—Rh2—Cl2	86.71 (8)
C9—Rh2—Cl1	90.5 (3)	Rh2—Cl1—Rh1	86.88 (8)
C10—Rh2—Cl1	176.7 (3)	Rh2—Cl2—Rh1	86.71 (7)
C9—Rh2—Cl2	176.8 (3)		

H atoms were constrained during the refinement with C—H distances in the range 0.93–0.97 \AA .

Data collection: *COLLECT* (Nonius, 1998); cell refinement: *DENZO* (Otwinowski & Minor, 1997); data reduction: *DENZO*;

program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *PLATON* (Spek, 1990); software used to prepare material for publication: *SHELXCIF-97* (Sheldrick, 1997).

Supplementary data for this paper are available from the IUCr electronic archives (Reference: GD1124). Services for accessing these data are described at the back of the journal.

References

- Boeyens, J. C. A., Denner, L., Orchard, S. W., Rencken, I. & Rose, B. G. (1986). *S. Afr. J. Chem.* **39**, 229–232.
- Corradi, E., Masciocchi, N., Palyi, G., Ugo, R., Vizi-Orosz, A., Zucchi, C. & Sironi, A. (1997). *J. Chem. Soc. Dalton Trans.* pp. 4651–4655.
- Dahl, L. F., Martell, C. & Wampler, D. L. (1961). *J. Am. Chem. Soc.* **83**, 1761–1762.
- De Ridder, D. J. A. & Imhoff, P. (1994). *Acta Cryst.* **C50**, 1569–1572.
- Ibers, J. A. & Snyder, R. G. (1962). *Acta Cryst.* **15**, 923–930.
- Nonius (1998). *COLLECT*. Nonius BV, Delft, The Netherlands.
- Otwinowski, Z. & Minor, W. (1997). *Methods Enzymol.* **276**, 307–326.
- Sheldrick, G. M. (1997). *SHELXL97*, *SHELXS97* and *SHELXCIF-97*. University of Göttingen, Germany.
- Spek, A. L. (1990). *Acta Cryst.* **A46**, C-34.
- Walz, L. & Scheer, P. (1991). *Acta Cryst.* **C47**, 640–641.